

Min Xue and Shi-Xiong Liu*

Department of Chemistry, Fuzhou University,
Fuzhou, Fujian 350002, People's Republic of
ChinaCorrespondence e-mail:
shixiongliu@yahoo.com

Key indicators

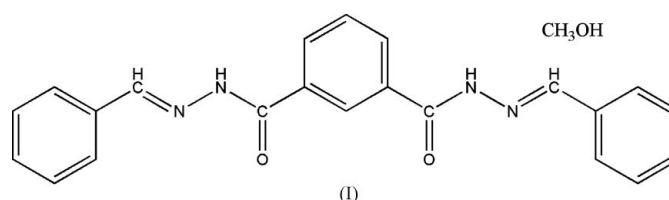
Single-crystal X-ray study
 $T = 293$ K
Mean $\sigma(\text{C}-\text{C}) = 0.003$ Å
 R factor = 0.047
 wR factor = 0.155
Data-to-parameter ratio = 18.3For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.2',2'-Dibenzylideneisophthalohydrazide
methanol solvate

The title compound, $\text{C}_{22}\text{H}_{18}\text{N}_4\text{O}_2 \cdot \text{CH}_3\text{OH}$, was synthesized by the reaction of benzaldehyde and isophthaloyl hydrazine in methanol. The molecule is non-planar, the dihedral angles between the pairs of aromatic rings being 13.2 (1), 27.0 (1) and 18.4 (1)°. The hydrazide molecules are linked *via* hydrogen bonds into a chain along the c axis.

Received 5 January 2006
Accepted 18 January 2006

Comment

The chemistry of aroylhydrazone compounds has received increasing attention because the hydrazone group is strongly coordinated to many metal atoms and aroylhydrazone compounds possess widespread applications in the treatment of tuberculosis. They also exhibit fungicidal activity (Edwards *et al.*, 1975; Zhi *et al.*, 2003; Yang & Pan, 2004). We report here the synthesis and crystal structure of the title compound, (I), obtained by the condensation of benzaldehyde with isophthaloyl hydrazine.



The molecular structure of (I) is shown in Fig. 1. The title molecule is non-planar. The dihedral angle between rings C1–C6 and C9–C14 is 13.2 (1)°, between rings C1–C6 and C17–C22 is 27.0 (1)° and between rings C9–C14 and C17–C22 is 18.4 (1)°. Similar $\text{C}=\text{O}$ distances (Table 1) have been

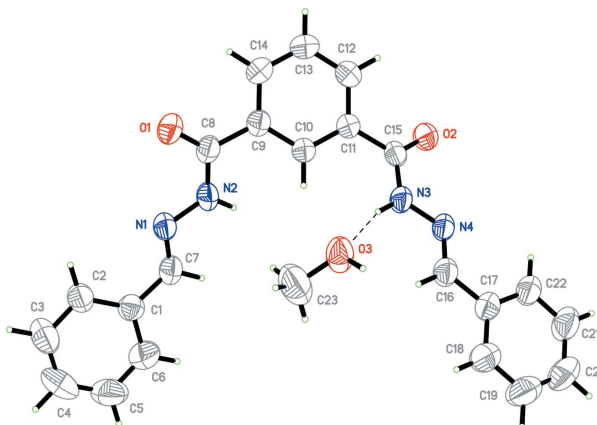


Figure 1

The molecular structure of (I), showing the atom-numbering scheme and 50% probability displacement ellipsoids. The dashed line represents a hydrogen bond.

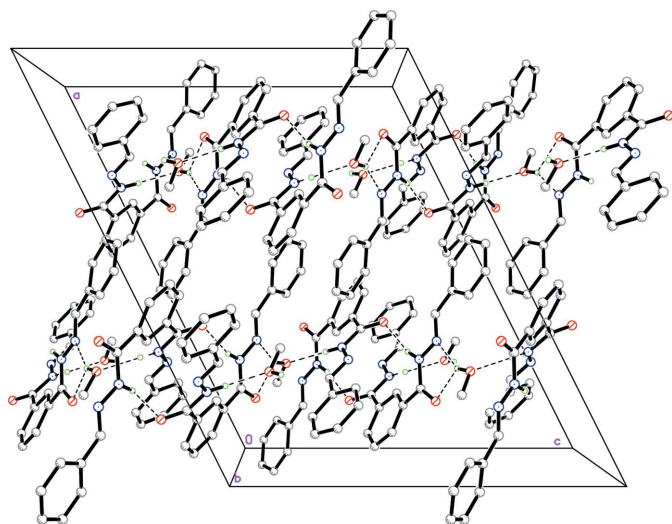


Figure 2
The packing of (I), showing the intermolecular hydrogen bonds as dashed lines. H atoms not involved in hydrogen bonding have been omitted.

observed in many hydrazone compounds. The N1—C7 and N4—C16 bond lengths are close to the value of 1.280 (5) Å found for the imine bond length in *p*-dimethylenedioxybenzaldehyde 2,4-dinitrobenzoylhydrazone (Wang *et al.*, 2004) and shorter than the value of 1.337 (2) Å found for the C—N single bond in the 1:1 complex of 1-phenyl-3-methyl-4-benzoyl-5-pyrazolone and nicotinoylhydrazine (Liu *et al.*, 2001). The N1—N2 and N3—N4 bond lengths are close to the values 1.3794 (19) and 1.388 (2) Å in *p*-dimethylenedioxybenzaldehyde benzoylhydrazone (Fun *et al.*, 1997), indicating that a partially conjugated system operates in this hydrazone.

There are four intermolecular hydrogen bonds in the crystal structure (Table 2), which link adjacent molecules to form a chain. The H atom on O3 engages in binding to atoms N1ⁱⁱ and O1ⁱⁱ simultaneously.

Experimental

Isophthaloyl hydrazine (17.8 mmol, 3.45 g) was dissolved in anhydrous methanol (50 ml), and benzaldehyde (35.7 mmol, 3.65 ml) was added. The mixture was refluxed for 3 h and the resulting precipitate was collected by filtration and washed with methanol and diethyl ether. The product (0.37 g) was dissolved in methanol (15 ml) and CH₂Cl₂ (15 ml), and kept at room temperature for 20 d to obtain colourless single crystals.

Crystal data

C₂₃H₂₂N₄O₃
M_r = 402.45
Monoclinic, C2/c
a = 20.168 (4) Å
b = 14.737 (3) Å
c = 16.357 (3) Å
β = 116.54 (3)°
V = 4349.0 (15) Å³
Z = 8

D_x = 1.229 Mg m⁻³
Mo Kα radiation
Cell parameters from 4999 reflections
θ = 1.8–27.5°
μ = 0.08 mm⁻¹
T = 293 (2) K
Prism, colorless
0.55 × 0.42 × 0.36 mm

Data collection

Rigaku Weissenberg IP diffractometer
ω scans
Absorption correction: multi-scan (TEXRAY; Molecular Structure Corporation, 1999)
T_{min} = 0.778, T_{max} = 0.970
20697 measured reflections

4999 independent reflections
3243 reflections with I > 2σ(I)
R_{int} = 0.031
θ_{max} = 27.5°
h = 0 → 26
k = -19 → 19
l = -21 → 19

Refinement

Refinement on F²
R[F² > 2σ(F²)] = 0.047
wR(F²) = 0.155
S = 1.04
4999 reflections
273 parameters

H-atom parameters constrained
w = 1/[σ²(F_o²) + (0.0901P)²]
where P = (F_o² + 2F_c²)/3
(Δ/σ)_{max} < 0.001
Δρ_{max} = 0.20 e Å⁻³
Δρ_{min} = -0.23 e Å⁻³

Table 1
Selected geometric parameters (Å, °).

| | | | |
|----------------|-------------|--------------|-------------|
| C1—C7 | 1.454 (2) | C15—O2 | 1.2305 (18) |
| C7—N1 | 1.270 (2) | C15—N3 | 1.348 (2) |
| C8—O1 | 1.2219 (17) | C16—N4 | 1.272 (2) |
| C8—N2 | 1.351 (2) | C16—C17 | 1.463 (2) |
| C8—C9 | 1.499 (2) | N1—N2 | 1.3838 (18) |
| C11—C15 | 1.497 (2) | N3—N4 | 1.3837 (16) |
| O1—C8—N2 | 122.96 (15) | C7—N1—N2 | 114.50 (13) |
| O2—C15—N3 | 122.84 (14) | C16—N4—N3 | 115.12 (13) |
| O1—C8—C9—C10 | 153.04 (15) | O1—C8—N2—N1 | -4.5 (2) |
| C10—C11—C15—O2 | 151.34 (15) | O2—C15—N3—N4 | 0.7 (2) |

Table 2
Hydrogen-bond geometry (Å, °).

| D—H···A | D—H | H···A | D···A | D—H···A |
|---------------------------|------|-------|-------------|---------|
| N2—H2A···O2 ⁱ | 0.86 | 2.17 | 3.0048 (17) | 164 |
| O3—H3C···N1 ⁱⁱ | 0.89 | 2.28 | 3.0954 (18) | 153 |
| O3—H3C···O1 ⁱⁱ | 0.89 | 2.30 | 2.9550 (19) | 131 |
| N3—H3A···O3 | 0.86 | 2.05 | 2.8848 (19) | 163 |

Symmetry codes: (i) $-x + \frac{1}{2}, -y + \frac{1}{2}, -z$; (ii) $-x + \frac{1}{2}, y + \frac{1}{2}, -z + \frac{1}{2}$.

Atoms H3A and H3C were located in difference Fourier maps, but were then allowed to ride on N3 and O3, with N—H = 0.86 Å and O—H = 0.89 Å. The other H atoms were placed in idealized positions (aromatic C—H = 0.93 Å, methanol C—H = 0.96 Å and N—H = 0.86 Å) and were refined using a riding model, with U_{iso}(H) = 1.5U_{eq}(C).

Data collection: TEXRAY (Molecular Structure Corporation, 1999); cell refinement: TEXRAY; data reduction: TEXSAN (Molecular Structure Corporation, 1999); program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEX (McArdle, 1995); software used to prepare material for publication: SHELXL97-2 (Sheldrick, 1997).

The authors are grateful for financial support from the National Natural Science Foundation of China (Nos. 20431010 and 20171012).

References

- Edwards, E. I., Epton, R. & Marr, G. (1975). *J. Organomet. Chem.* **85**, C23–C25.
- Fun, H.-K., Lu, Z.-L., Duan, C.-Y., Tian, Y.-P., You, X.-Z., Gong, X.-Y. & Guo, Y.-M. (1997). *Acta Cryst.* **C53**, 1454–1455.
- Liu, L., Jia, D.-Z., Qiao, Y.-M. & Yu, K.-B. (2001). *Acta Chim. Sin.* **59**, 1495–1501.
- McArdle, P. (1995). *J. Appl. Cryst.* **28**, 65.
- Molecular Structure Corporation (1999). *TEXRAY* (Version 1.10) and *TEXSAN* (Version 1.10). MSC, 9009 New Trails Drive, The Woodlands, TX 77381-5209, USA.
- Sheldrick, G. M. (1997). *SHELXL97*, *SHELXS97* and *SHELXL97-2*. University of Göttingen, Germany.
- Wang, J.-L., Jia, Y.-J., Miao, F.-M. & Li, A.-X. (2004). *Chin. J. Org. Chem.* **24**, 41–49.
- Yang, J.-G. & Pan, F.-Y. (2004). *Acta Cryst.* **E60**, o2009–o2010.
- Zhi, J. F., Bin, Z., Su, H. W. & Zheng, M. L. (2003). *Chin. J. Appl. Chem.* **20**, 365–367.